SVISCISVAS

Application Note

May 1, 2022

Keywords or phrases: CSAR, drug docking and screening, structure-activity screening

Generating Reliable Kinetic Data for Protein-Ligand Interactions

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Abstract

The Community Structure Activity Resource (CSAR) is hosted by the University of Michigan and is focused on developing a database of protein-ligand structures and their associated affinities through participation of the scientific community. The goal of this project, funded by the National Institute of General Medical Sciences (NIGMS) is to increase the amount of high-quality data publicly available for development, validation and benchmarking of ligand docking and screening software. Datasets are collected with the aim of improving scoring functions and developing new docking algorithms.

The Octet[®] R series has dramatically increased the rate at which data can be produced for CSAR and has become an essential technology for advancing this important initiative.

Introduction

The Community Structure Activity Resource (CSAR, www.csardock.org) group is developing a database of high quality protein-ligand structures and the corresponding binding affinities. The data will be provided from in-house experiments and community collaborations. The proteins are generally well-studied structures that have been targeted in drug discovery projects. Current projects include CDK2, LpxC, and urokinase. The drug-like ligands for each of these targets consist of several series of compounds with a wide range of affinities.

The Octet[®] R2 system has dramatically increased the data we can produce in a short time, turning it into the workhorse of the CSAR project. Note: The Octet[®] R2 system referred to in the application note has been replaced by the Octet[®] R8 system. To learn more about the Octet[®] platform, visit www.sartorius.com.

Approach

Current docking and scoring models rely on experimental data, which is typically incomplete: crystal structures of protein-ligand complex may exist, but affinity data is not available. Affinity data for a series of compounds might be published, but the crystal data may be poor quality or incomplete. Variations between experimental conditions can create differences between published data. Without high-quality datasets, training these models becomes difficult if not impossible. Our goal is to create complete high-quality data sets to assist computational scientists with their goals.

To populate the affinity database, three label-free methods are used: the Octet[®] R2 system, a thermal denaturation assay, and isothermal titration calorimetry (ITC). These produce both complementary and overlapping data. The methods use dissimilar technologies to independently produce affinity values (K_D) with exactly the same compound, protein, and buffer. This helps to reduce experimental error and gives modelers more confidence in the accuracy of the data.

Methods

The Octet[®] R2 system determines rate constants, k_a and $k_{a'}$ which are used to calculate a $K_{\rm D}$, defined as k_a/k_a . We determine $K_{\rm D}$ for six different compounds in two hours using six concentrations for each compound and 0.1 mg of protein. After scouting runs to determine optimal concentrations of compound, pre-made compound plates allow us to complete up to four runs in a single day. This throughput has dramatically increased the data we can produce in a short time, turning the Octet[®] R2 system into the workhorse of the CSAR project.

The thermal denaturation assay is performed on the Thermofluor[®] platform. It is a high-throughput method that detects changes in fluorescence as the protein is heated and melts. Ligand added to the protein stabilizes it and increases the melting temperature. The difference between the melting temperatures of the protein and the proteinligand complexes allows calculation of an affinity value. A run takes about two hours, and uses a comparable amount of protein and compound to the Octet® R2 system. The K_D values, however, must be calculated using the Van't Hof equation. Some of the variables used for the equation are difficult to obtain and must be estimated, therefore the affinity value calculated from the melting temperature difference is an estimation at best. The dye used to create the fluorescent signal also has the potential to interact with the protein in unforeseen ways. An article recently published indicated that the 1,8 ANS dye binds allosterically to CDK2, causing a conformation change. This unexpected challenge invalidates our current Thermofluor data and requires us to repeat experiments with a new dye.

The third method we use is ITC. It has a long history of being a reliable method and gives affinity data very similar to that produced by the Octet[®] R2 system. ITC has a number of advantages and disadvantages. It is the slowest of the three methods, but unique in that it can provide accurate enthalpy of binding values. Each run uses several milligrams of protein that must be saturated with 2 to 3 times the molar equivalent of compound. In addition, it can measure only a very limited range of affinities. For the CDK2 project, the affinities of only six of 27 compounds could be used, due to issues with ligand solubility and affinity range. The enthalpy alone was determined for an additional four compounds. This data can be combined with the affinity data from the Octet[®] R2 system to determine the Gibbs free energy and entropy of binding. Table 1: Data obtained for CDK2 binding to compounds of molecular weight ranging from 193 Da to 509 Da. All experiments were performed at T = 298K.

Compound	Octet [®] R2 system (average)			ITC (average	ITC (average)		ΔG , Calculated average	
	<i>К</i> _D (М)	<i>k</i> _a (1/Ms)	k _d (1/s)	$K_{\rm D}$ (M)	∆H (kcal/mol)	ITC	Octet [®] R2 system	
CS1	3.32E-05	3.57E+04	1.09E+00		-7.52		-6.11	
CS2	1.84E-06	2.22E+05	3.81E-01	4.23E-06	-8.76	-7.33		
CS3	8.08E-07	3.35E+05	2.59E-01		-17.70		-8.31	
CS4	2.12E-05	7.19E+04	1.29E+00	6.03E-06	-7.95	-7.12		
CS9	5.83E-05	2.35E+04	2.58E-01		-12.70		-6.50	
CS10	9.10E-07	7.97E+04	6.66E-02				-8.24	
CS11	2.80E-07	2.81E+05	8.40E-02	1.41E-06	-11.75	-7.98		
CS13	9.65E-07	6.95E+04	3.94E-02				-8.21	
CS14	1.43E-06	6.43E+04	7.70E-02				-7.97	
CS15	5.57E-07	1.13E+05	3.67E-02				-9.00	
CS16	7.73E-08	2.94E+05	1.69E-02				-9.70	
CS17	5.39E-08	1.22E+05	3.40E-03				-10.16	
CS18	9.61E-07	1.00E+05	8.23E-02				-8.21	
CS19	2.29E-07	1.75E+05	2.01E-02				-9.06	
CS20	7.77E-07	8.59E+04	5.62E-02				-8.33	
CS241	2.87E-07	175033.3	4.97E-02				-8.92	
CS242	5.28E-07	3.21E+04	1.73E-02		-6.57		-8.56	
CS244	1.79E-07	9.73E+04	1.28E-02				-9.20	
CS245	2.29E-07	9.72E+04	8.20E-03				-9.06	
CS246	5.26E-07	1.14E+05	5.92E-02	1.94E-07	-14.99	-9.15		
CS247	2.26E-08	9.18E+04	2.30E-03				-10.40	
CS248	1.06E-06	7.84E+04	7.97E-02				-8.15	
CS260	9.86E-08	1.32E+05	1.33E-02	1.79E-07	-10.70	-9.20		
CS261	2.51E-07	3.97E+04	1.00E-02	1.28E-06	-3.39	-8.04		
CS262	4.73E-08	6.49E+04	2.61E-03				-10.00	

Conclusion

Consistent results are the most important aspect of using the Octet® R2 system. Data must be consistent between runs as well as between different methods. The Octet® R2 system data have been very consistent with ITC. In cases where it has been possible to use all three methods on a compound, the data produced by each method are similar (as a general rule, an affinity within 3-fold of one determined via another method is considered identical) and this is shown in Table 1. Some of the inconsistencies in our data may be due to compound solubility issues. The Octet® R2 system requires far lower ligand concentrations than ITC, and thus is less affected by compound solubility issues. Thermal denaturation assays are often used for compounds with low solubility, but still require saturation of the protein. The Octet® R2 system can produce usable data at concentrations 10–100 fold lower than ITC and Thermofluor, helping ensure accurate compound concentration in solution.

The results presented in this application note were included in the first data set for the CSAR database. Additional data sets for CDK2-Cyclin A and Lpxc can be found at www.csardock.org. CSAR will continue to create high-quality protein-ligand structures and affinity datasets to aid the computational chemistry community.

Thermofluor is a trademark of Johnson and Johnson. The ITC measurements were made with the Nano ITC Low Volume system from TA Instruments.

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